

## Master's program Entrance examination topics

## Chemistry

- The periodic table of the elements. Isotopes, electronic structure, the Bohr-Sommerfeld model. Bonding models, hybridization, VSEPR theory, molecular orbital theory, properties of chemical bonds, primary and secondary bonds.
- 2. States of matter, gases, kinetic gas theory, diffusion, gas laws, fluids, surface tension, cohesive forces. Solid state, crystal structures, changes in states, phase diagrams.
- 3. Chemical equilibria, acid-base reactions and theories. Mixtures of gases, fluids, mixing, phase equilibria in mixtures. Colligative properties of dilute solutions.
- 4. Thermodynamics: laws, heat, entropy, enthalpy (reaction enthalpy), Gibbs free energy (equilibrium processes), electrochemistry (reactions on a boundary surface).
- 5. Aliphatic hydrocarbons: alkanes and derivatives, stereoisomers, the Cahn-Ingold-Prelog convention. Newman- and Fischer-projection, aromatic hydrocarbons, aromaticity.
- 6. Classification of organic reactions, nucleophilicity and electrophilicity, energy profile of reactions. Reactivity: inductive, steric and mesomeric effects. Mechanisms of reactions. Additions and eliminations.
- 7. Oxo compounds: aldehydes and ketones, tautomerism, carboxylic acids and their derivatives, carbohydrates and heterocyclic compounds.
- 8. Amino acids and proteins, protein folding, primary, secondary and tertiary structure, cooperativity
- 9. Nucleosides, nucleotides and nucleic acids, their structure and biological function.
- 10. Enzymes: fundamentals of enzyme catalysis, basic mechanisms, examples.
- 11. Bioenergetics: role and production of ATP, completion of energetically unfavorable reactions in the cells.
- 12. Lipids, biological membranes and transport mechanisms.

## Recommended literature

James E. House: Inorganic chemistry. Elsevier, 2008.

Leroy G. Wade, Jan W. Simek: *Organic Chemistry*. Pearson, 2016.

Jeremy M. Berg, John L. Tymoczko, Gregory J. Gatto, Jr., Lubert Stryer: *Biochemistry*.